

Preliminary communication

ALKYNYL THIOETHER COMPLEXES OF CHROMIUM, MOLYBDENUM AND TUNGSTEN

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Summary

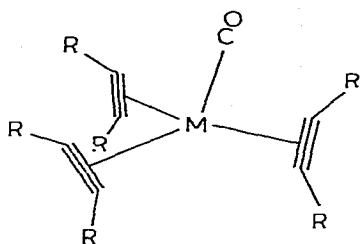
Treatment of $\text{MeSC}\equiv\text{CSMe}$ (L) with $[\text{M}(\text{CO})_3(\text{MeCN})_3]$ ($\text{M} = \text{Cr}, \text{Mo}, \text{W}$) has given $[\text{M}(\text{CO})\text{L}_3]$ ($\text{M} = \text{Mo}, \text{W}$) and $[\text{Cr}(\text{CO})_2\text{L}_2]$ and the reactions of these π -acetylene complexes with ditertiary phosphines are described, it is suggested that trimerisation of L on chromium proceeds through a $[4+2]$ π cycloaddition.

Very few π -acetylene complexes of the Group VI metals are known [1]. In particular, no simple π -acetylene derivative of $\text{Cr}(\text{CO})_6$ has been described, although $[(\eta\text{-arene})\text{Cr}(\text{CO})_2(\text{R}_2\text{C}_2)]$ and $[\eta\text{-C}_5\text{H}_5\text{Cr}(\text{CO})(\text{NO})(\text{R}_2\text{C}_2)]$ are well known [2]. Most of the work on acetylene complexes of the transition metals has concerned a relatively restricted range of alkyl and aryl acetylenes, particularly where the organic group is electron attracting (e.g. CF_3 , CO_2R). Apart from some recent work on alkynyl phosphines [3] and silanes [4], very little attention has been given to the organometallic chemistry of other hetero-substituted acetylenes. Despite the rapid development of the chemistry of alkynyl ethers and thioethers and their extensive application in organic synthesis [5], almost nothing [6] is known of their reactions with transition metals.

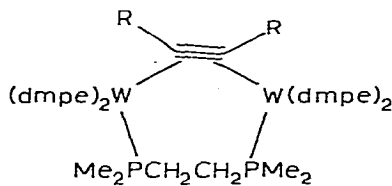
Reaction between $[\text{M}(\text{CO})_3(\text{MeCN})_3]$ ($\text{M} = \text{Mo}, \text{W}$) and 2,5-dithia-3-hexyne, $\text{MeSC}\equiv\text{CSMe}$ (L), in ethanol produces the yellow crystalline complexes, $[\text{M}(\text{CO})\text{L}_3]$ in high yield (90-95%). These new compounds, which have been fully characterised by microanalysis and spectroscopy (for $\text{M} = \text{W}$, m.p. 448-452 K (dec.); $\nu(\text{CO})$ 2081 (CH_2Cl_2 solution) cm^{-1} ; $\delta(\underline{\text{C}}\text{H}_3)$ 2.78, $\delta(\text{CH}_3)$ 23.0 ppm; m/e 566 (100%)), most probably have structure I ($\text{M} = \text{Mo}, \text{W}$) similar to that [1] of $[\text{W}(\text{CO})(\text{Ph}_2\text{C}_2)_3]$; there is no evidence of sulphur coordination. The addition of the ditertiary phosphine $\text{Me}_2\text{PCH}_2\text{CH}_2\text{PMe}_2$ (dmpe) to an equimolar quantity of $[\text{W}(\text{CO})\text{L}_3]$ in CH_2Cl_2 solution

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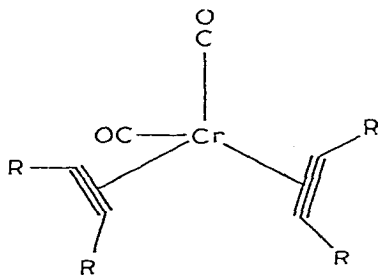
at room temperature rapidly produces the salmon pink complex $[\text{W}(\text{CO})(\text{dmpe})\text{L}_2]$ ($\nu(\text{CO}) 1903 \text{ cm}^{-1}$; $\lambda_{\text{max}} 488 \text{ nm}$). With excess dmpe, an insoluble white solid complex $[\text{W}_2\text{L}(\text{dmpe})_5]$ is formed which may have the structure indicated by II. Hexakis(methylthio)benzene is not formed in these displacement reactions. Under similar conditions there is no detectable reaction between $\text{Ph}_2\text{PCH}_2\text{CH}_2\text{PPh}_2$ and $[\text{W}(\text{CO})\text{L}_3]$. A similar distinction in reactivity arises in the case of $[\text{W}(\text{CO})(\text{Ph}_2\text{C}_2)]$.



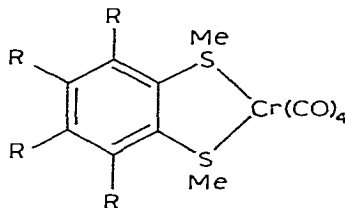
(I)



(II)



(III)



(IV)

The dropwise addition of a cooled solution of L in ether to $[\text{Cr}(\text{CO})_3(\text{MeCN})_3]$ stirred in ether at 195 K and warming to allow reaction at 230 K in the dark, followed by removal of the volatile materials in the dark gives violet needles of $[\text{Cr}(\text{CO})_2\text{L}_2]$ (III) (Found: C, 35.0; H, 3.5; Cr, 14.8; S, 37.0. $\text{C}_{10}\text{H}_{12}\text{CrO}_2\text{S}_4$ calcd.: C, 34.9; H, 3.4; Cr, 15.0; S, 37.2%). This photosensitive compound is easily converted both in the solid state and, more rapidly, in solution to a mixture of products of which by far the major constituent is a brown crystalline solid complex *cis*- $[\text{Cr}(\text{CO})_4(\text{L}_3)]$ (IV) ($\nu(\text{CO}) 2023\text{m}, 1910\text{s}, 1879\text{s}$ (CH_2Cl_2 solution) cm^{-1}), in which the chelating ligand is hexakis(methylthio)benzene obtained by trimerisation of the acetylene, L. Destruction of IV gives pure $\text{C}_6(\text{SMe})_6$ (*m/e* 354). Addition of dmpe to III gave IV and, subsequently, *cis*- $[\text{Cr}(\text{CO})_4(\text{dmpe})]$.

The surprising contrast in the constitution of I and III, with the observation that trimerisation of L occurs from III but not from I under comparable

conditions, may suggest that trimerisation occurs as a result of a [4 + 2] π -cycloaddition, as in the case [7] of $[\eta\text{-C}_5\text{H}_5\text{Nb}(\text{CO})(\text{Ph}_4\text{C}_4)(\text{Ph}_2\text{C}_2)]$, rather than as a result of a [2 + 2 + 2] π -cycloaddition [8].

The reactions of L and other alkynyl thioethers with a variety of organo-metallic compounds have been examined and the results will be reported in due course.

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